

## Emission and Transport of Air Pollutants

### Lecture 4

#### Calculation of emission and air pollution

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## Emission and concentration of pollution

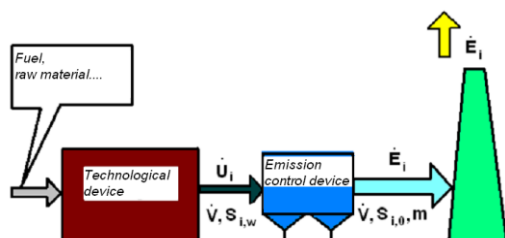
$\dot{U}_i$  - emission before control emission device – directly  
from source of emission, kg/h, kg/Mg of product

$\dot{E}_i$  - emission of pollution, kg/h

$S_{iw}, S_{i0}$  - concentration of pollution in gas before and  
after control emission device, mg/m<sup>3</sup>,  
mg/m<sup>3</sup><sub>u</sub>

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## Emission and concentration of pollution



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## Pollution concentration

Concentration before emission control device, mg/m<sup>3</sup>

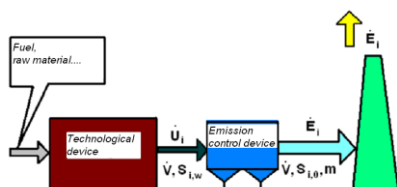
$$S_{iw} = \frac{\dot{U}_i}{\dot{V}}$$

Concentration after emission control device, mg/m<sup>3</sup>

$$S_{i0} = \frac{\dot{E}_i}{\dot{V}}$$

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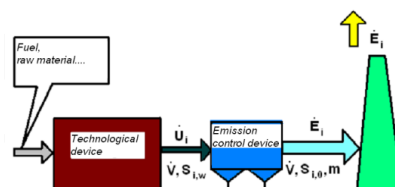
## Emission



When  $\dot{E}_i = \dot{U}_i$  ?

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## Emission



$\dot{E}_i = \dot{V} \cdot S_i$

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## Emission from installation

(Dz.U.nr 260, poz.2181 z zm.)

Emission standard from instalation relate to:

- **dry gas stream** ,  $\text{m}^3_u/\text{h}$   $\dot{V}_{ss} = n''_{ss} \cdot \dot{P} \cdot 22,42$
- **standard conditions**:  $T = 273 \text{ K}$ ,  $p = 101,3 \text{ kPa}$
- **strict define oxygen content** in gas exhaust

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## Strict define oxygen content in gas exhaust

- **3%** - in case of combustion of liquid and gaseous fuels, as well as co-combustion of liquid fuels with waste,
- **6%** - in case of:
  - combustion of solid fuels: hard coal, brown coal, coke and biomass,
  - co-combustion of solid fuels with waste and biomass and co-combustion of waste with biomass,
  - estimation of concentration of heavy metals, dioxins and furans in exhaust gas from waste combustion process,
- **11%** - in case of waste combustion and co-combustion, when thermal power from combustion of hazardous waste is over 40% of nominal thermal power, with strictly precised exceptions,
- **15%** - in case of combustion of liquid and gaseous fuels in turbines.

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## Recalculation of substances' concentration in exhaust gas for standard oxygen amount

$$S_{iu}^{[O_2]=X\%} = S_{ipom} \frac{21 - [O_2]_u}{21 - [O_2]_{pom}}$$

where:

$S_{iu}^{[O_2]=X\%}$  – concentration of  $i$  substance in conventional conditions for standard oxygen concentration  $X\%$ ,  $\text{mg}/\text{m}^3_u$

$S_{ipom}$  – calculated, or measured concentration of  $i$  substance in conventional conditions,  $\text{mg}/\text{m}^3_u$

$[O_2]_u$  – conventional (standard) oxygen concentration in dry exhaust gas, %

$[O_2]_{pom}$  – calculated, or measured oxygen concentration in dry exhaust gas, %.

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## Recalculation of substances' concentration in exhaust gas for standard oxygen amount

In case of combustion with oxidizer of defined oxygen concentration (different than in the air):

$$S_{iu}^{[O_2]=X\%} = S_{ipom} \frac{[O_2]_{utl} - [O_2]_u}{[O_2]_{utl} - [O_2]_{pom}}$$

where:

$[O_2]_{utl}$  – volumetric oxygen part in oxidizer

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## Recalculation of substances' concentration in exhaust gas for standard oxygen amount in polish regulations (Dz. U. nr 260, poz. 2181)

### § 4.2

2. Stężenie substancji w gazach odlotowych apro-wadza się do standardowej zawartości tlenu w gazach odlotowych, obliczając według następującego wzoru:

$$E_1 = \frac{21 - O_1}{21 - O_2} \times E_2$$

gdzie:

$E_1$  – oznacza stężenie substancji w gazach odlotowych przy standardowej zawartości tlenu w gazach odlotowych,

$E_2$  – oznacza stężenie substancji w gazach odlotowych (zmierzony albo obliczony),

$O_1$  – oznacza standardową zawartość tlenu w gazach odlotowych, wyrażoną w procentach,

$O_2$  – oznacza zawartość tlenu w gazach odlotowych, wyrażoną w procentach (zmierzoną albo obliczoną).

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## Permissible emission – emission standards from installations (Dz. U. nr 260, poz. 2181)

Permissible emission of  $i$  substance introduced from installation to the atmosphere

- Permissible concentration at standard oxygen concentration:  $D_i$ ,  $\text{mg}/\text{m}^3_u$

- Permissible emission:  $\dot{E}_{id}$ ,  $\text{kg}/\text{h}$ ,  $\text{Mg}/\text{rok}$

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### Minimal pollutant reduction ratio

Minimal pollutant reduction ratio of  $i$  substance introduced to the atmosphere in concentration above to the permissible one:

$$\eta_i = \frac{S_{iu}^{[O_2]=X\%} - D_i}{S_{iu}^{[O_2]=X\%}} \cdot 100\%$$

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### Permissible emission amount

Permissible emission amount of  $i$  pollutant introduced to the atmosphere, kg/h

$$\dot{E}_{id} = \frac{D_i}{S_{iu}^{[O_2]=X\%}} \cdot \dot{E}_i$$

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### Index method of emission calculation

Types of measurements:

- Continuous – installations of thermal power not less than 100 MW, where is realized process of combustion and co-combustion of waste and emission of VOC
- Periodical – transport, small boilers, etc.

**Emission index** is the mass of pollutant, expressed in **kg**, which is emitted from installation during combustion of **1 Mg of solid fuel, 1 m<sup>3</sup> of liquid fuel or 10<sup>6</sup> m<sup>3</sup> of gaseous fuel**

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### Index method of emission calculation

$$\dot{U}_i = \dot{P} \cdot w_i$$

where:

$\dot{U}_i$  - emission of  $i$  substance, kg/h

$\dot{P}$  - fuel stream, Mg/h

$w_i$  - emission index: emission of  $i$  substance (kg) from combustion of fuel unit (1 Mg) in particular technology, kg/Mg

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### Index method of emission calculation for solid fuel

Emission of SO<sub>2</sub>

$$\dot{E}_{SO_2} = \dot{P} \frac{w_{SO_2} \cdot s \cdot (100 - \eta)}{100}$$

Emission of NO<sub>2</sub>, CO, CO<sub>2</sub>, benzo(a)piren

$$\dot{E}_i = \dot{P} \frac{w_i \cdot (100 - \eta)}{100}$$

where

s- sulfur content in fuel

w<sub>i</sub> – emission index, kg/Mg of fuel

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### Permissible air pollution

Permissible levels of substances in atmospheric air are defined for different **areas of country**, for different **time periods**, and for **standard condition (293 K, 101,3 kPa)**

- Permissible one hour averaged concentration – D<sub>1</sub>, µg/m<sup>3</sup>

- Permissible annual averaged concentration – D<sub>a</sub>, µg/m<sup>3</sup>

- Permissible dust-fall on area – D<sub>p</sub>, g/(m<sup>2</sup>rok)

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### Permissible air pollution

Permissible pollution of  $\text{NO}_x$ , as well as emission from installation is recalculated on amount of  $\text{NO}_2$ .

Recalculation of NO concentration to  $\text{NO}_2$

$$S_{\text{NO}_2} = 1,5332 S_{\text{NO}} \quad \text{mg/m}^3_u$$

Total concentration of  $\text{NO}_2$  in gas

$$S_{\text{NO}_2} = S_{\text{NO}_2} + 1,5332 S_{\text{NO}} \quad \text{mg/m}^3_u$$

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### Permissible air pollution (Dz.U. nr 16 poz. 87)

Reference amount of pollutants concentration, or permissible levels are fulfilled when frequency of exceeding of reference amount by one-hour averaged concentration is no longer than **0,247%** of year in case of  $\text{SO}_2$ , and no longer than **0,2%** or the **rest of pollutants**.

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### Exam questions

1. Measurements in spa area showed, that concentration of NO and  $\text{NO}_2$  is, respectively, 0,1 and 0,05  $\text{mg/m}^3$ . Measurements were carried out in temperature 293 K and pressure 101,3 kPa. Does the permissible  $\text{NO}_x$  concentration ( $D_1 = 200 \text{ mg/m}^3$ ) is exceeded?
2. Recalculate 50 ppm<sub>v</sub> benzene concentration into  $\text{mg/m}^3$ .
3. Calculate  $\text{SO}_2$  emission from combustion of 50 Mg/h of coal containing 1,25 mass % of sulphur. 80% of sulphur is combustible one, and desulphurization efficiency of exhaust gas is 90%.
4. What is index method of emission calculation?

### Exam questions

5. Calculate conventional CO concentration from waste combustion process. Measured CO concentration is  $150 \text{ mg/m}^3_u$ , and  $\text{O}_2$  content in exhaust gas is 8%.
6. Calculate minimal efficiency of  $\text{SO}_2$  purifying installation in case where 30-minutes averaged permissible concentration is  $D_{\text{SO}_2/30} = 200 \text{ mg/m}^3_u$ . Concentration of  $\text{SO}_2$  is  $S_{\text{SO}_2} = 400 \text{ mg/m}^3_u$  for 11% oxygen content.
7. Calculate permissible CO emission from waste combustion process, when day averaged concentration is  $D_{\text{CO}} = 50 \text{ mg/m}^3_u$ , measured concentration is  $S_{\text{CO}} = 80 \text{ mg/m}^3_u$ , and measured emission is  $E_{\text{CO}} = 80 \text{ kg/h}$ . What conclusion can be drawn from calculations?